

Solid Phase Epitaxy of Rutile-type GeO₂ on TiO₂ (001)

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Abstract

Rutile-type GeO₂ (r-GeO₂) is an ultra-wide bandgap oxide semiconductor, which attracts growing interest because of its ambipolar dopability and high carrier mobility predicted by recent first principles calculations. While epitaxial thin films of r-GeO₂ have been synthesized using various vapor phase epitaxy techniques, desorption of germanium suboxide limits the growth conditions in a narrow range. Here, we demonstrated solid phase epitaxy (SPE) of r-GeO₂ thin films on TiO₂ (001) substrates. Amorphous GeO₂ thin films fabricated using pulsed laser deposition at ≤ 250 °C were crystallized by post-deposition annealing at 700 °C. X-ray diffraction and cross-sectional scanning transmission electron microscopy analysis revealed that the amorphous GeO₂ thin films epitaxially crystallized to r-GeO₂ phase without any impurity phases. The epitaxial crystallization of r-GeO₂ was significantly promoted by introducing a seed layer of a rutile Ge_xSn_{1-x}O₂ epitaxial thin film, which coherently grown on the TiO₂ substrate, probably due to chemical interactions between the amorphous film and the seed layer. The crystallinity of the SPE-grown r-GeO₂ was comparable to that synthesized via vapor phase epitaxy, indicating that the SPE is an alternative route for synthesizing r-GeO₂ epitaxial thin films.

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1 Ultra-wide bandgap semiconductors, of which bandgap (E_g) is larger than conventional
2 wide bandgap semiconductors such as GaN and SiC, have been extensively studied for power
3 electronics and ultraviolet optoelectronics applications owing to their high electric breakdown
4 field and bandgap energy corresponding to ultraviolet light.¹⁻³ Among ultra-wide bandgap
5 semiconductors, rutile-type GeO₂ (r-GeO₂) ($E_g \sim 4.7$ eV)⁴ attracts growing interest because of its
6 advantageous features for power electronics applications, such as ambipolar dopability and high
7 carrier mobility, which are predicted by first-principles calculations,⁵⁻⁸ as well as high thermal
8 conductivity.⁹

9 To experimentally validate these unique properties expected for r-GeO₂ and to fabricate
10 a device structure, synthesis of r-GeO₂ in both bulk single crystal and epitaxial thin film forms is
11 indispensable. Bulk single crystals of r-GeO₂ have been synthesized since the 1960s by chemical
12 vapor transport and solution processes.^{6,10-13} Recently, n-type dopability with high carrier electron
13 concentration of up to 2.2×10^{20} cm⁻³ was reported for Sb-doped r-GeO₂ bulk single crystals
14 synthesized via top seeded solution growth.¹³ On the other hand, there is no report on successful
15 doping in r-GeO₂ epitaxial thin films, although various vapor phase deposition techniques such
16 as molecular beam epitaxy (MBE),¹⁴ pulsed laser deposition (PLD),^{15,16} mist chemical vapor
17 deposition,^{17,18} metal-organic chemical vapor deposition,¹⁹ and low-pressure chemical vapor
18 deposition²⁰ have been proposed after the first demonstration using MBE.¹⁴

19 There are two main difficulties in synthesizing r-GeO₂ thin films. The first one is
20 thermodynamic stability competing against the α -quartz phase and the amorphous phase, of which
21 formation energies are comparable to that of r-GeO₂ phase.^{6,14} The second one is the desorption
22 of germanium suboxide (GeO),²¹⁻²³ which limits the growth parameters (e.g. growth temperature
23 and partial oxygen pressure during the deposition) in a narrow range.¹⁴ As a result, thin films of
24 r-GeO₂ frequently suffered from the coexistence of the other phases and/or poorly crystallized
25 regions.^{15,24,25}

26 In this study, we demonstrated solid phase epitaxy (SPE) of r-GeO₂ thin films on TiO₂

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(001) substrates as an alternative route for synthesizing r-GeO₂ epitaxial thin films. In the SPE process, amorphous GeO₂ films were crystallized epitaxially from the film/substrate interface by post-deposition annealing. In the previous studies on the crystallization of amorphous GeO₂ into r-GeO₂, the synthesized films were polycrystalline,^{26,27} some of which are mixed phase with α -quartz phase or Ge. It was revealed that the SPE of the r-GeO₂ was significantly enhanced by introducing an epitaxial rutile Ge_xSn_{1-x}O₂ (GSO) film as a seed layer, on which epitaxial crystallization of r-GeO₂ was promoted probably due to chemical interactions between the amorphous film and the seed layer.

The SPE of r-GeO₂ was conducted by the following three steps: (Step 1) A ~10-15 nm-thick (001)-oriented rutile Ge_xSn_{1-x}O₂ (GSO) thin film (seed layer) was epitaxially grown on a rutile TiO₂ (001) substrate to promote the epitaxial crystallization as discussed later. (Step 2) An amorphous GeO₂ film was deposited on the GSO seed layer at low temperature. (Step 3) The amorphous GeO₂ film was crystallized by annealing inside the growth chamber without exposure to air to avoid degradation of the amorphous by humidity.

The GSO seed layer (Ge content $x \sim 0.4$) and the amorphous GeO₂ film were fabricated by PLD. The growth condition of the GSO seed layer was described elsewhere.¹⁸ The amorphous GeO₂ thin films were deposited at substrate temperature of 250 °C by ablating a ceramic pellet of GeO₂ using a KrF excimer laser (Coherent, COMPex pro 102). The pulse repetition rate and the laser fluence were set at 2 Hz and 1-2 J·cm⁻²·pulse⁻¹, respectively, to control the deposition rate at ~100 nm/h. The base pressure in the growth chamber was lower than 1×10^{-7} Torr. Partial O₂ pressure was kept at 1×10^{-4} Torr during the deposition of amorphous GeO₂ thin films and the post-annealing for crystallization, where the supplied O₂ gas was activated into radicals by an electron cyclotron resonance plasma source (Tectra, Gen2). It is noted that GeO₂ film deposited at substrate temperature of 600 °C was totally evaporated during the deposition (Supplemental Fig. S1).

1 Crystal structure and density of the GeO₂ thin films were evaluated by X-ray diffraction
2 (XRD) and X-ray reflection (XRR) measurements, respectively, using a four-circle diffractometer
3 (Bruker AXS, D8 DISCOVER). The structure of the amorphous GeO₂ thin film was evaluated by
4 Raman spectroscopy (Renishaw, inVia Qontor). The microscopic structures and the chemical
5 composition distribution in the r-GeO₂ thin film were investigated by cross-sectional scanning
6 transmission electron microscopy (STEM) (JEOL, JEM-ARM200F-B, 200 kV) with a detector
7 for energy dispersive X-ray spectroscopy (EDS) (Thermo Fisher Scientific, NSS). The specimen
8 for STEM observation was prepared by using a focused ion beam system (Hitachi High-Tech,
9 NB5000). The surface morphologies of the films were observed by atomic force microscopy
10 (AFM) (SII-nanotechnology, SPI4000 with SPA400). Thicknesses of the thin films were
11 examined by XRR measurement or using a stylus surface profiler (Kosaka Laboratory Ltd.,
12 Surfcomer ET200 A).

13
14 Figure 1a shows θ - 2θ XRD patterns of the GeO₂ thin films deposited at 250 °C on the
15 GSO seed layer and annealed at various temperature. It is noted that the GeO₂ film deposited at
16 600 °C was totally evaporated during the deposition (Supplemental Fig. S1). The GeO₂ thin film
17 annealed at 600 °C showed diffraction peaks only from the GSO seed layer and the TiO₂ substrate,
18 indicating that the GeO₂ film was amorphous. On the other hand, a clear 002 diffraction peak of
19 r-GeO₂ was observed for the film annealed at 700 and 750 °C without the diffraction peaks from
20 the α -quartz type GeO₂ nor the r-GeO₂ with different crystallographic orientations. XRD
21 reciprocal space map (RSM) of the r-GeO₂ films around the 112 diffraction (Fig. 1b) and phi scan
22 measurement (Fig. 1c) verified the epitaxial growth of (001)-oriented r-GeO₂ without any
23 rotational domains. Epitaxial growth of the r-GeO₂ was also confirmed at partial O₂ pressure of
24 1×10^{-3} Torr without the plasma activation (Supplemental Fig. S2). The GeO₂ film annealed at
25 800 °C showed diffraction peaks only from the substrates (Fig. 1a) indicating the evaporation of
26 the r-GeO₂ film and the GSO seed layer during the annealing. Partial evaporation of GeO₂ layer

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1 was also suggested in the films annealed at 700 and 750 °C (Supplemental Table S3). Based on
2 these results, controlling the temperature of the post deposition annealing at 700-750 °C is
3 required for the SPE of r-GeO₂. Although this temperature range is not wider than that of vapor
4 phase epitaxy, the high temperature limit could be extended by suppressing the evaporation of
5 GeO₂ under more oxidative conditions as reported in the previous studies on the crystallization of
6 amorphous GeO₂ thin films.^{26,27} It is also speculated that slightly introducing O vacancies in the
7 amorphous GeO₂ thin film, which kinetically enhances atomic diffusion, enables the
8 crystallization at lower temperature than 700 °C.²⁸ Off-stoichiometry of the amorphous GeO₂ thin
9 film was not observed by Raman spectroscopy (Supplemental Figure S4).

10 The lattice constants of the r-GeO₂ epitaxial thin films determined from the RSM ($a =$
11 $4.410 \text{ \AA}$, $c = 2.859 \text{ \AA}$) were close to those reported for bulk r-GeO₂ ($a = 4.3975 \text{ \AA}$, $c = 2.8625$
12 $\text{\AA}$)²⁹, indicating that the r-GeO₂ film was almost relaxed. The full width of half maximum
13 (FWHM) of the rocking curve of 002 diffraction was 0.365° (Fig. 1d), which is comparable to the
14 values reported for the r-GeO₂ epitaxial thin films synthesized from vapor phase using PLD.^{15,16}

15 For the SPE of r-GeO₂ on TiO₂, introducing the GSO seed layer is crucial. Most of the
16 amorphous films deposited directly on the TiO₂ (001) substrate did not crystallize by the annealing
17 at 700 °C. Even in the case of epitaxially crystallized film (Fig. 1a), the crystallinity of the film
18 was lower than those grown on the GSO seed layer: FWHM of the rocking curve of 002
19 diffraction increased from 0.365° to 1.03° (Fig. 1d). Notably, the GSO seed layer coherently grew
20 on the TiO₂ substrate without strain relaxation ($a = 4.594 \text{ \AA}$, $c = 3.088 \text{ \AA}$) as shown in the RSM
21 (Fig. 1b), meaning that lattice mismatch between the r-GeO₂ thin film and the TiO₂ substrate ($a =$
22 $4.594 \text{ \AA}$, $c = 2.959 \text{ \AA}$) was not reduced by the insertion of the GSO layer. This is in stark contrast
23 to the previous reports on the epitaxial growth of r-GeO₂,^{14,18} where the introduction of a GSO
24 buffer layer stabilizes rutile structure through the reduction of the lattice mismatch between the r-
25 GeO₂ thin film and the substrate. This result suggested that the epitaxial nucleation of the r-GeO₂
26 at the film/substrate interface was dominated not only by the elastic energy originating from the

1 lattice mismatch but also by the chemical interactions between the film and the substrate.

2 The microscopic structure of the r-GeO₂ thin film was further investigated by cross-
3 sectional STEM observation. Fig. 2a shows the cross-sectional high-angular annular dark field
4 (HAADF) images of the r-GeO₂ epitaxial thin film grown on the GSO seed layer. Selected area
5 electron diffraction patterns (Figs. 2b, c) show diffraction spots only from the (001)-oriented r-
6 GeO₂, confirming the epitaxial crystallization of r-GeO₂. Z-contrast in the HAADF-STEM image
7 and the results of EDS mapping (Fig. 3) show a sharp and flat interface between the r-GeO₂ thin
8 film and the GSO seed layer verifying that no interdiffusion occurred at the interface during the
9 SPE process. The HAADF-STEM images of the r-GeO₂ film show almost no spatial
10 inhomogeneity in its contrast except for the small regions probably corresponding to dislocations
11 and grain boundaries. Considering the large difference in atomic density between the amorphous
12 and r-GeO₂, the amorphous GeO₂ film was fully crystallized into the rutile phase from the
13 film/seed layer interface to the film surface.

14 The θ -2 θ peak and the rocking curve of the GSO seed layer became broader after the
15 post deposition annealing (Supplemental Figure S5), which indicates degradation of crystallinity
16 during the annealing. Based on the results of STEM-EDS mapping (Fig. 3), this degradation did
17 not originate from the interdiffusion between the GSO and the GeO₂, which was also supported
18 by a negligible shift in the XRD θ -2 θ peak of the seed layer after the annealing (Fig. S2). Because
19 the GSO seed layer annealed without the amorphous GeO₂ layer also showed a broadening of the
20 XRD peaks (Fig. S2), we speculate that the deterioration of the seed layer was caused by the
21 thermal instability of GSO.³⁰

22 Figure 4 shows AFM images of the amorphous GeO₂ and the SPE-grown r-GeO₂ thin
23 films (thickness of ~120 nm). While the surface of the amorphous GeO₂ (Fig. 4a) was smooth
24 (root-mean-square roughness $r_{\text{RMS}} = 0.24$ nm), grain-like morphology appeared after the
25 crystallization (Fig. 4b). Surface roughness of the SPE-grown r-GeO₂ thin film ($r_{\text{RMS}} = 3.94$ nm)
26 was larger than the amorphous GeO₂ ($r_{\text{RMS}} = 0.24$ nm), although the value was comparable to or

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1 slightly better than the values reported for r-GeO₂ thin films grown by vapor phase epitaxy.^{16,19,20}
 2 The increase of the roughness is, at least in part, due to the evaporation of GeO during the
 3 annealing (Supplemental Table S3), although it might also be attributable to the large difference
 4 in density between the amorphous GeO₂ thin films (~4.2 g cm⁻³, slightly larger than bulk GeO₂
 5 glass³¹⁻³³) and the r-GeO₂ epitaxial thin films (~6.2 g cm⁻³, almost the same as bulk r-GeO₂ of
 6 6.27 g cm⁻³).

7

8 In summary, (001)-oriented r-GeO₂ epitaxial thin films were grown on TiO₂ (001) substrates
 9 by the SPE of amorphous GeO₂ thin films deposited using PLD. The epitaxial crystallization of
 10 the amorphous GeO₂ film was significantly enhanced by the insertion of an epitaxial GSO seed
 11 layer. The crystallinity and surface roughness of the SPE-grown r-GeO₂ thin films were almost
 12 comparable to the r-GeO₂ epitaxial thin films synthesized by vapor phase epitaxy. These results
 13 demonstrate that the SPE is a useful technique with a potential for enlarging the process window
 14 of r-GeO₂, which will promote future studies on its physical properties and device applications.

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Supplementary Material

See the supplementary Material for details on the XRD pattern of the GeO₂ deposited at 600 °C, XRD patterns of the r-GeO₂ thin films crystallized without the plasma activation of O₂ gas, thickness of the GeO₂ layer before and after crystallization, Raman spectrum of the amorphous GeO₂ layer, and XRD patterns of the GSO seed layer after the post deposition annealing.

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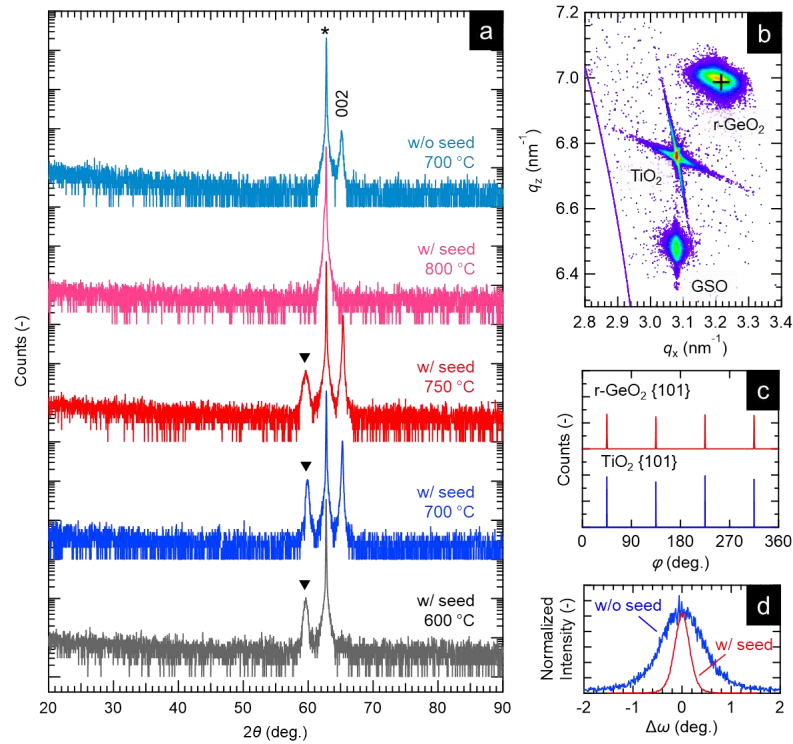


Figure 1. (a) θ -2 θ XRD patterns of the GeO₂/GSO (001)/TiO₂ (001) (w/ seed) after the post deposition annealing at 600, 700, 750 and 800 °C. Diffraction pattern of the GeO₂/TiO₂ (001) (w/o seed) crystallized at 700 °C is also plotted for comparison. Asterisk and triangles denote the diffraction peaks of the TiO₂ substrate and the GSO seed layer, respectively. (b) RSM around 112 diffraction peak of the r-GeO₂ (001)/GSO (001)/TiO₂ (001) crystallized at 700 °C. Cross indicates the peak position calculated from the lattice constants of bulk r-GeO₂. (c) Phi-scan of the 101 diffraction for the r-GeO₂ (001)/GSO (001)/TiO₂ (001). (d) Rocking curves of 002 diffraction for r-GeO₂ (001) epitaxial thin films crystallized at 700 °C with and without the GSO seed layer.

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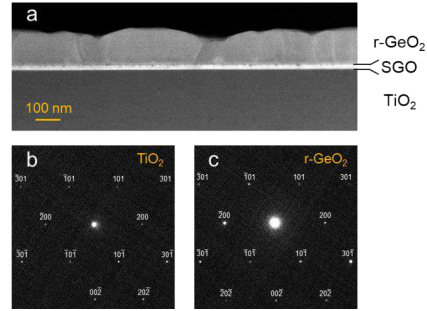


Figure 2. (a) Cross-sectional HAADF STEM image and (b, c) SAED patterns of the r-GeO₂ (001)/SGO (001)/TiO₂ (001) synthesized by the SPE process. Diffraction pattern (b) and (c) were obtained for the TiO₂ substrate and the r-GeO₂ film, respectively along the [010] zone axis.

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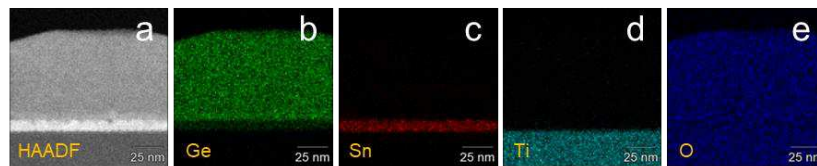


Figure 3. (a) STEM-HAADF image of the EDS mapping area of the r-GeO₂ (001)/GSO (001)/TiO₂ (001). (b-e) Elemental maps constructed from the EDS signals of (b) Ge K-line, (c) Sn L-line, (d) Ti K-line, and (e) O K-line.

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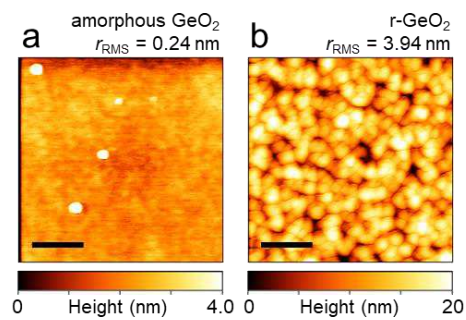


Figure 4. AFM image of (a) amorphous GeO₂ and (b) r-GeO₂ epitaxial thin film. The scale bar is 500 nm.